



## Methane China High Purity Wholesale Cylinder Gas Colorless CH<sub>4</sub> Gas

Our Product Introduction

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### Basic Information

- Place of Origin: China
- Brand Name: CMC
- Certification: COA
- Model Number: CH<sub>4</sub>
- Minimum Order Quantity: 1kg
- Price: US \$12/kg
- Packaging Details: Cylinder/Tank
- Delivery Time: 15 days
- Payment Terms: L/C, T/T
- Supply Ability: 5000kg/month



## Methane Gas

### Product Specification

- Product Name: Methane Gas
- Valve: Cga350
- Boiling Point: -161.5 °C
- Melting Point: -182.5 °C
- Cylinder Pressure: 15MPa/20MPa
- Cylinder Standard: DOT/ISO/GB
- Transport Package: Sea Transportation
- Specification: 8L/40L/47L/50L
- Trademark: CMC
- Origin: Suzhou, China
- CAS No.: 74-82-8
- Formula: CH<sub>4</sub>
- EINECS: 200-812-7
- Constituent: Industrial Pure Air
- Grade Standard: Industrial Grade



### More Images



## Product Description

### Methane China High Purity Wholesale Cylinder Gas Colorless CH4 Gas

Methane gas (CH<sub>4</sub>) is a colorless, odorless, and flammable gas. It is the primary component of natural gas and is one of the most abundant greenhouse gases in the Earth's atmosphere. Here are some key points about methane gas:

**Chemical Composition:** Methane is composed of one carbon atom bonded to four hydrogen atoms (CH<sub>4</sub>).

**Properties:** Methane possesses several important properties:

**Flammability:** Methane is highly flammable and can ignite in the presence of an ignition source, such as a spark or flame.

**Greenhouse Gas:** Methane is a potent greenhouse gas, meaning it has a strong ability to trap heat in the Earth's atmosphere. It contributes to climate change and global warming.

**Odorless:** Pure methane gas is odorless. However, in commercial and residential natural gas supplies, an odorant called mercaptan is added to give it a distinctive smell for easy detection of gas leaks.

**Occurrence and Sources:** Methane can be found in various natural and human-made sources:

**Natural Sources:** Methane is emitted naturally from sources such as wetlands, oceans, and the digestive processes of animals (e.g., cows, termites). It is also produced during the decomposition of organic matter in landfills.

**Fossil Fuels:** Methane is the primary component of natural gas, which is extracted from underground reservoirs and used as a fuel for heating, cooking, and electricity generation. It is also released during the extraction, production, and distribution of coal, oil, and natural gas.

**Agriculture:** Livestock farming, especially cattle, produces significant amounts of methane through enteric fermentation (digestive processes) and manure management.

**Energy and Waste Management:** Methane can be produced from the anaerobic decomposition of organic waste in wastewater treatment plants, landfills, and anaerobic digesters.

**Environmental Impact:** Methane is a critical greenhouse gas with significant environmental impacts:

**Climate Change:** Methane has a much higher global warming potential (GWP) than carbon dioxide (CO<sub>2</sub>) over a 20-year period. It contributes to the greenhouse effect and plays a role in climate change.

**Air Quality:** Methane itself is not harmful to human health, but its combustion can produce pollutants such as nitrogen oxides (NO<sub>x</sub>) and carbon monoxide (CO), which can have adverse effects on air quality and human health.

**Mitigation and Control:** Reducing methane emissions is crucial for addressing climate change and improving air quality. Some strategies to mitigate methane emissions include:

**Methane Capture and Utilization:** Implementing technologies to capture methane emissions from landfills, wastewater treatment plants, and agricultural operations and using it as a valuable energy source.

**Improved Agricultural Practices:** Implementing practices to reduce enteric fermentation in livestock, such as improved diets and waste management.

**Leak Detection and Repair:** Regular monitoring and maintenance of natural gas infrastructure to identify and repair leaks promptly.

**Renewable Energy:** Shifting from fossil fuels to renewable energy sources helps reduce overall methane emissions associated with energy production.

Methane gas is a significant contributor to climate change and global warming. Efforts to reduce methane emissions are crucial for mitigating these impacts and transitioning to a more sustainable energy and waste management system.

#### Basic Info.

Molecular Weight	16.043	Density	0.717G/L
Melting Point	-182.5°C	Boiling Point	-161.5°C
Appearance	Colorless,Odorless	Un No.	1971
DOT Class	2.1	Valve	CGA350
Cylinder Standard	DOT/ISO/GB	Cylinder Pressure	15Mpa/20Mpa
Transport Package	40L/47L/50L	Specification	99.9%,99.99%,99.999%
Trademark	CMC	Origin	China
HS Code	27112900	Production Capacity	20000m³/Year

#### Specification:

Specification	Company Standard
CH <sub>4</sub>	≥ 99.999%
N <sub>2</sub>	≤ 2.0 ppm
O <sub>2</sub> +AR	≤ 1.0 ppm
H <sub>2</sub>	≤ 1.0 ppm
CO	≤ 0.5 ppm
CO <sub>2</sub>	≤ 0.5 ppm
Ne	≤ 1.0 ppm
CH <sub>4</sub>	≤ 0.5 ppm
Moisture	≤ 0.5 ppm

#### Detailed Photos









## Company Profile





Shanghai Kemike Chemical Co., Ltd is staffed by trained personnel, combine many years experience in Gas industry .We supply cylinder gas, electronic gas, etc ., and the gas holder, panel, valves and fittings and other equipment, parts and engineering services to our customers in China and worldwide; The products are involved in various industrial fields, such as semiconductor chip, solar cell, LED, TFT-LCD, optical fiber, glass, laser, medicine , etc., Our mission is to partner with our global customers to provide support, solutions and quality products that are innovative, reliable, and safe. Our products mainly include: H<sub>2</sub>, O<sub>2</sub>, N<sub>2</sub>, Ar, CO<sub>2</sub>, propane, acetylene, helium, laser mixed gas, SiH<sub>4</sub>, SiH<sub>2</sub>Cl<sub>2</sub>, SiHCl<sub>3</sub>, SiCl<sub>4</sub>, NH<sub>3</sub>, CF<sub>4</sub>, NF<sub>3</sub>, SF<sub>6</sub>, HCL, N<sub>2</sub>O, doping mixed gas (TMB, PH<sub>3</sub>, B<sub>2</sub>H<sub>6</sub>) and other electronic gases.

SiCl <sub>4</sub>	NH <sub>3</sub>	NH <sub>3</sub>	CH <sub>3</sub> F	SiH <sub>4</sub>	Kr	H <sub>2</sub> S	WF <sub>6</sub>	F <sub>6</sub> +Cl <sub>2</sub>
4MS	C <sub>3</sub> F <sub>8</sub>	C <sub>3</sub> F <sub>8</sub>	TEOS	CH <sub>4</sub>	PH <sub>3</sub>	SF <sub>6</sub>	C <sub>2</sub>	HCl+Ne
CF <sub>4</sub>	C <sub>4</sub> F <sub>8</sub>	SiH <sub>2</sub>						TMB+H <sub>2</sub>
SiF <sub>4</sub>	C <sub>3</sub> H <sub>8</sub>	Cl <sub>2</sub>						He +As
BBr <sub>3</sub>	C <sub>3</sub> H <sub>6</sub>	DCE						Ge+Se
POCl <sub>3</sub>	N <sub>2</sub>	SO <sub>2</sub>						D+B
BCl <sub>3</sub>	D <sub>2</sub>	CO <sub>2</sub>						CO+NO
SiHCl <sub>3</sub>	CH <sub>2</sub> F <sub>2</sub>	HF						Ar+O <sub>2</sub>
TMAI	DMZn	DEZn						Xe+NO
AsH <sub>3</sub>	C <sub>2</sub> H <sub>4</sub>	C <sub>2</sub> H <sub>2</sub>	HBr	COS	Ar+O <sub>2</sub>			
GeH <sub>4</sub>	C <sub>2</sub> H <sub>6</sub>	B <sub>2</sub> H <sub>6</sub>	H <sub>2</sub> Se	GeCl <sub>4</sub>	Xe+NO			





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